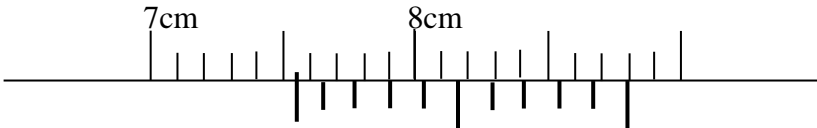


**PHYSICS PAPER 232/1
FORM FOUR
MARKING SCHEME**

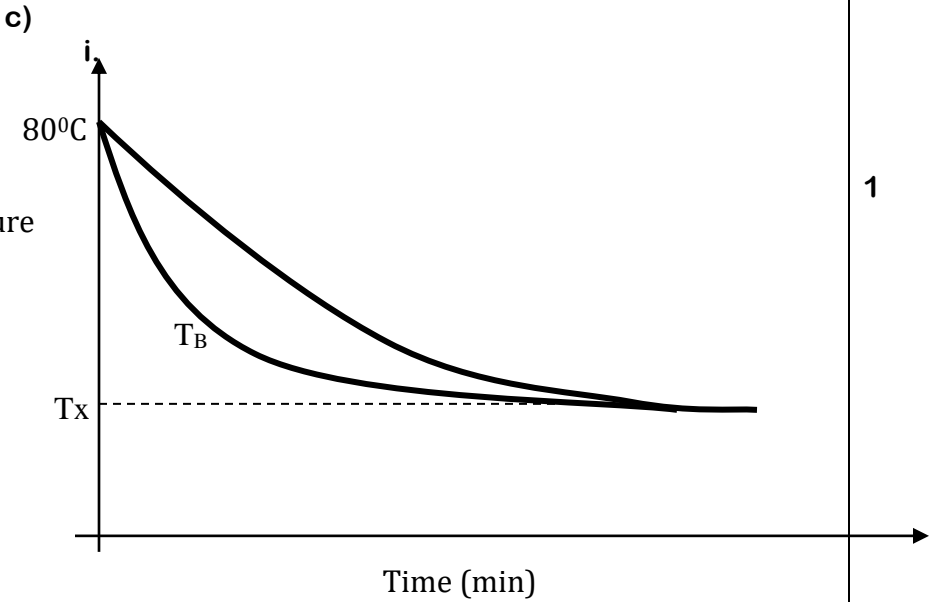
SECTION A (25 marks)			
1.	<p>Reading + error = Actual</p> <p>$R = 7.57 - 0.04 = 7.53\text{cm}$</p> 	<p>1</p> <p>1</p> <p>1</p>	<p>Calculation of R</p> <p>3rd vernier scale aligning</p> <p>Complete vernier scale inserted correctly</p>
2.	<p>a. B</p> <p>large area of force application implying less pressure is exerted.</p> <p>b.</p> $\text{pressure of trapped air} = P_{atm} - (P_{\text{due to water}} + P_{\text{mercury}})$ $P_{air} = \left(\frac{75}{100} \times 13600 \times 10\right) - \left(\frac{58}{100} \times 1000 \times 10 + \frac{20}{100} \times 13600 \times 10\right)$ <p>$P_{air} = 102,000 - (5800 + 27200)$</p> <p>$= 102,000 - 33000$</p> <p>$= 69,000\text{Pa}$</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p>	
3.	Uniform acceleration	1	
4.	<p>Zero acceleration (constant velocity)</p> <p>(W = F + U) the body has attained a terminal velocity</p>	<p>1</p> <p>1</p>	
5.	When the temperature increases, the <u>gas molecules gain more kinetic energy and move faster</u> ; the <u>rate at which they collide increase</u> .	<p>1</p> <p>1</p>	
6.	<u>More stable or stability increases</u> when the candle melts the <u>position of COG lowers</u>	<p>1</p> <p>1</p>	
7.	Diffusion involves movement of matter and it can only happen when in tiny particles.	1	
8.	Due to <u>the resultant downward cohesive force</u> on the molecules on the surface of liquid, <u>the net force on molecule inside the liquid is zero</u> .	<p>1</p> <p>1</p>	

9.	$RD \text{ of paraffin} = \frac{\text{upthrust in paraffin}}{\text{upthrust in water}}$ $\frac{0.3 - 0.2}{0.3 - 0.15} = \frac{0.1}{0.15} = 0.67$ $\rho = 0.67 \times 1000 = 670 \text{ kg/m}^3$	1 1 1	
10.	<p>A</p> <p>When air is blown through A, pressure inside reduces but in B remains the same. The higher atmospheric pressure outside in A makes it collapse.</p> <p>ALT. In A there is pressure difference between inside and outside but in B there is no pressure difference.</p>	1 1	
11.	<p>Conduction involves passing of heat energy from one molecule to the next through vibration. Convection involves heated molecules moving up on becoming lighter and are replaced by colder molecules (convectional currents)</p>	2	tied

SECTION B (55 MARKS)

12.	<p>a)</p> <ul style="list-style-type: none"> i. Pointer ii. Note and record the initial length of the spring before loading Record the new length after loading Get the difference iii. Weight of the mass/stretching force iv. <ul style="list-style-type: none"> • Assuming elastic limit of the spring is not exceeded • Note and record the extension of the spring and the corresponding value of the weight of the mass/stretching force. • Increase the mass at intervals and note and record several values of extension and weight of masses in a table. • Plot a graph of stretching force against extension. • Note the nature of the graph (it will be a straight line through the origin). <p>b)</p> <p>One spring = 2cm K=F/e</p> <p>F = 150/1000 x 10 = 1.5N</p>	1 1 1 1 1 1 1 1 1	
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	$K = 1.5/2$ $= 0.75\text{N/cm or } 75\text{N/m}$	1 1	
		11	
13.	<p>a)</p> $F \times 0.8 = (90 + 20) \times 10 \times 0.2$ $F = \frac{110 \times 10 \times 0.2}{0.8} = 275\text{N}$ <p>b)</p> $VR = \frac{\text{effort dist}}{\text{load dist}} = \frac{0.8}{0.2}$ $= 4$ <p>c)</p> $MA = \frac{\text{load}}{\text{effort}} = \frac{900}{275}$ $= 3.273$ <p>d)</p> $\text{eff.} = \frac{MA}{VR} \times 100$ $\text{eff.} = \frac{3.273}{4} \times 100$ $= 81.82\%$ <p>e) Some energy is used to lift the weight of the machine (20kg)</p>	1 1 1 1 1 1 1 1 1 1 1	
		12	
14.	<p>a)</p> <ul style="list-style-type: none"> i. The bulb of the thermometer is dipped in the boiling water. Should be in the steam. ii. Impuriites in water would raise its boiling point hence the mark would be higher. iii. To ensure that the steam is at normal/standard atmospheric pressure by allowing excess steam out. <p>b)</p> $100^\circ\text{C} \Rightarrow 11 - 3 = 8\text{cm}$ $80^\circ\text{C} \Rightarrow 80/100 \times 8 = 6.4\text{cm}$	1 1 1 1 1	

	<p>Length = 3 + 6.4 = 9.4cm</p> <p>c)</p>  <p>ii. Room temperature</p> <p>iii. Radiation</p> <p>d) Nature of the material</p> <p>Temperature difference between the ends of the conductor</p> <p>e)</p> <p>Large interparticle distances</p> <p>They do not have free electrons.</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p>	
		12	
15.	<p>a) Specific latent heat of fusion of ice is the amount of heat required to change unit mass of ice completely to water at constant temperature.</p> <p>b)</p> <p>i. Heat lost by steam = $mLv + mc\Delta T$</p> $\frac{60}{1000} \times 2.26 \times 10^6 + \frac{60}{1000} \times 4200 \times (100 - 40)$ <p>= 150,720J</p> <p>ii. $m_i L_f + m_i c \Delta T + m_w c \Delta T + m_c c_c \Delta T$</p> $3.34 \times 10^5 \times m_i + m_i \times 4200 \times 40 + 0.1 \times 4200 \times 40 + 0.05 \times 400 \times 40$	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p>	

	<p>$3.34 \times 10^5 \times m_i + 168,000m_i + 16800 + 800$</p> <p>$502,000m_i + 17,600$</p> <p>iii.</p> <p>$502,000m_i + 17,600 = 150,720$</p> $m_i = \frac{150,720 - 17,600}{502,000}$ <p>$= 0.265\text{kg or } 265\text{g}$</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p>	
		10	
16.	<p>a) The direction is continuously changing. This implies change in velocity hence acceleration</p> <p>b)</p> $\omega = 2\pi f = 2 \times 3.142 \times 6 = 37.704\text{rads /s}$ <p>$a = v^2/r = r \omega^2 = 37.704^2 \times 0.6$</p> <p>$a = 852.955\text{m/s}$</p> <p>c)</p> <p>i. $\frac{50-0}{25-0} = 20\text{N/kg}$</p> <p>ii. $\frac{P}{m} = \text{slope}$</p> <p>$P = m \times \text{slope} = 20 \times 0.2 = 4.0\text{N}$</p> <p>iii. P represent Centripetal force</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>2</p> <p>1</p> <p>1</p> <p>1</p>	
	TOTAL	55	
SECTION B (55 MARKS)			